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(54) **Biodegradable thermoplastic material**

(57) The invention relates to a biodegradable ther-

moplastic material comprising a natural polymer, a plasticizer and a clay.

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## Description

[0001] The invention relates to a biodegradable thermoplastic material and to a process for preparing it.

[0002] Typically, thermoplastic materials have been based on polymers that are synthetically prepared from hydrocarbon- or hydrocarbon-like raw materials. They thus find their origin in the petrochemical industry.

[0003] Meanwhile, these materials have become associated with environmental problems. Their production is relatively energy intensive and produces large quantities of carbon dioxide as a side product. Carbon dioxide is often believed to cause, or at least contribute to, global warming. Furthermore, when discarded as waste, these materials degrade only very slowly, which means that they constitute a prolonged burden on the environment.

[0004] In order to alleviate these problems, various alternative materials have been developed. These alternative materials have mostly been devised with an eye to their advantageous (bio)degradation profile, when discarded after use. In the context of the invention, a biodegradable material is a material which, in a biological environment, degrades in a relatively short period of time to compounds which are preferably water soluble and non-toxic. The degradation may proceed by hydrolysis under influence of light, air, water and/or micro-organisms.

[0005] Most of the known biodegradable thermoplastic materials are either also based on hydrocarbon sources, or based on natural raw materials (monomers) or even natural polymers, such as cellulose, starch, polylactic acid, keratin, and the like. These natural raw materials are, more or less intrinsically, biodegradable. Furthermore, they have the advantage that they originate from renewable sources and will therefore always be available. Natural polymers are, however, generally not thermoplastic. In order to achieve that property, the materials are typically processed (often extruded) with a plasticizer. Of course, the biodegradable properties of a suitable plasticizer are to be considered in its selection.

[0006] Unfortunately, in practice there are not many choices for the plasticizer. Usually, either water, urea, glycerol or a low aliphatic or aromatic ester is selected. Problems that are encountered are that these plasticizers either are insufficiently compatible with the biodegradable polymer, or may leach out of the product, which in its turn will become brittle and may even fall apart. This problem is particularly encountered in applications wherein the product is used in a humid or aqueous environment, i.e. when it is brought into contact with water. This disadvantage puts a serious limitation on the applications of the biodegradable thermoplastic material. It moreover means that the (mechanical) properties of the material deteriorate rather fast, making it unsuitable for use long before its biodegradation takes effect.

[0007] The present invention seeks to overcome the problems associated with the known biodegradable

thermoplastic materials from natural polymers. In particular, it is an object of the invention to provide a material, which is biodegradable and has good thermoplastic and mechanical properties, which material is highly compatible with biodegradable plasticizers. It is furthermore an object of the invention that the favorable properties of the biodegradable thermoplastic material remain apparent over a prolonged period of time, preferably at least until biodegradation affects said properties.

[0008] Surprisingly, it has been found that these objects can be reached by incorporating a specific clay into a biodegradable thermoplastic material. Accordingly, the invention relates to a biodegradable, thermoplastic material comprising a natural polymer, a plasticizer and a clay having a layered structure and a cation exchange capacity of from 30 to 250 milliequivalents per 100 gram.

[0009] Due to the presence of the clay, the plasticizer is substantially retained in the biodegradable thermoplastic material, thereby avoiding the problems with loss of plasticizer that were encountered with the known biodegradable thermoplastic materials. Hence, a material according to the invention has superior properties, and those properties are maintained over a prolonged period of time. In other words, the stability of a biodegradable thermoplastic material is significantly improved because of the presence of the clay. In accordance with the invention, a thermoplastic material is a material that is deformable upon increase of temperature.

[0010] In the prior art, a combination of a natural polymer, in this case a polysaccharide, and a clay has been disclosed in the German patent application 195 04 899. However, this combination is not a thermoplastic material as no plasticizer is present. Furthermore, the clay is used in combination with the polysaccharide merely in order to control the porosity of the material.

[0011] The natural polymer on which the present biodegradable thermoplastic material is based, may be any natural polymer that is conventionally used to serve as basis for a biodegradable thermoplastic material. Examples include carbohydrates (polysaccharides) and proteins. Particular good results have been obtained using starch, cellulose, chitosan, alginic acid, inulin, pectin, and derivatives thereof. Derivatives that may be used are for example esters, such as acetylated starch, or carboxymethylated cellulose, and ethers, such as hydroxypropylated starch.

[0012] In accordance with the invention, it has further been found that some of these natural polymers may be used without plasticizer, leading, in combination with the clay, to a biodegradable thermoplastic or thermosetting material. Natural polymers that have been found suitable for preparing a thermoplastic or thermosetting material in accordance with this embodiment are the above mentioned derivatives having a high degree of substitution (DS), typically at least 1. Specific examples include acetylated starch and hydroxypropylated cellulose.

[0013] A suitable plasticizer is a compound that is compatible with the other constituents of the material

and that is capable of imparting thermoplastic properties to the material. Suitable examples for the plasticizer include water, urea, glycerol, ethylene glycol, oligomers of ethylene glycol and mixtures thereof. Preferably, the plasticizer is used in an amount of 15 to 60 wt.%, more preferably of 25 to 45 wt.%, based on the weight of the thermoplastic material.

**[0014]** A material according to the invention comprises, as stated before, a clay having a layered structure. The clay may be of a natural or synthetic nature. Preferably, the clay has a large contact surface. Typically, the clay will be present in an amount of from 0.1 to 90 wt.%, preferably from 1 to 60 wt.%, based on the weight of the biodegradable thermoplastic material. The amount of clay relative to the amount of the natural polymer may be used as an instrument to adjust the (shear) viscosity of the biodegradable thermoplastic material. An increase in the amount of clay may lead to a decrease in the (shear) viscosity.

**[0015]** Very suitable are clay types based on layered silicates, such as layered phyllosilicate composed of magnesium and/or aluminum silicate layers which are each about 7-12 Å in thickness. Especially preferred are smectite-like clay minerals, such as montmorillonite, saponite, hectorite, fluorohectorite, beidellite, nontronite, vermiculite, halloysite and stevensite. These materials impart very favorable mechanical properties and a great heat resistance to a biodegradable thermoplastic material.

**[0016]** A suitable clay type has a cation exchange capacity of from 30 to 250 milliequivalents per 100 gram. When this capacity exceeds the above upper limit, it proves difficult to finely disperse the clay on a molecular level because of the strong mutual interaction of the clay layers. When the cation exchange capacity is lower than the above lower limit, it turns out that the clay is hard to modify, owing to the fact that the interaction with the natural polymer is small. Preferably, a clay having a cation exchange capacity of from 50 to 200 milliequivalents per 100 gram is used.

**[0017]** The biodegradable thermoplastic material may further comprise any conventional additives, such as pigments, stabilizers, processing aids, aroma substances, anti-fouling agents and the like.

**[0018]** Under specific circumstances, it may be desired that the material comprises a compatibilizing agent that assists in providing an optimal combination of the clay and the natural polymer. This possibility will be set forth in greater detail below in the discussion of the preparation of a material according to the invention, to which the invention also pertains.

**[0019]** In order to prepare a material according to the invention, first a suspension of the clay in water and/or another plasticizer is prepared. The suspension will typically contain between 1 and 80 wt.% of the clay. In order to ensure that a highly homogeneous suspension is achieved, the temperature of the water or the plasticizer is preferably between 25 and 60°C during the prepara-

tion of the suspension.

**[0020]** In a first embodiment, the exfoliated clay as prepared in a suspension as set forth above is mixed with the natural polymer under high shear, preferably in an extruder at elevated temperature. Preferred elevated temperatures are from 35 to 200°C, preferably from 60 to 180°C.

**[0021]** In a second embodiment, the clay in the suspension is ion exchanged with the natural polymer in a molar ratio of the reactive groups at the natural polymer, often OH or NH<sub>2</sub> groups, to the cation capacity of the clay ranging from 50:1 to 0.5:1. The degree of intercalation/exfoliation can be varied by a change and adjustment of the molecular weight of the natural polymer and the molar ratios of the reacting functional groups of said natural polymer. The thus obtained material may be used as such or in combination with pure natural polymers. In the latter case, it may be desired to carry out an extrusion as mentioned above.

**[0022]** In a third embodiment, particularly when the natural polymer contains hydrophobic groups, such as acetyl, alkylcarboxyl or arylcarboxyl ester groups, the exfoliated clay is first ion exchanged with a modifying agent, such as an onium ion. Suitable examples of onium ions include ammonium, phosphonium and sulfonium ions. This modification, which is known for incorporation of clay into other materials, has the goal to compatibilize the layers of the clay with the natural polymer. It utilizes suitable surfactants in an ion exchange reaction. The surfactant should have an onium functionality in addition to a functionality compatible with the natural polymer, such as one or more OH groups, COOH groups and the like. Preferably, the surfactant has from 6 to 16 carbon atoms. The surfactant will typically be employed in an amount of from 5 to 70 wt.%, preferably from 20 to 40 wt.%, with respect to the clay.

**[0023]** The above described biodegradable thermoplastic material is suitable for the production of various articles of manufacture. Examples of suitable applications of the material include coatings, films, packaging materials, carrier materials, construction materials and the like. The invention accordingly further relates to such articles of manufacture.

**[0024]** The invention will now be further elucidated by the following, non-restrictive examples.

## Examples

### Example 1:

**[0025]** A homogeneous mixture (4 g) of 40 w% clay and 60 w% demineralized water is blended in a twin-screw extruder at a temperature of 85-105 °C with a premixed powder consisting of 20 g dry native potato starch, 4.1 g glycerol, and 3.6 g water using a screw speed of 200 rpm. The resulting thermoplastic material shows a complete and homogeneous dispersion of the clay sheets within the starch matrix material (see Figure

1, which shows a TEM picture of the nanocomposite material as described in this example. A very good dispersion of the clay particles within the matrix material is seen.)

#### Example 2:

[0026] A suspension of a montmorillonite (1 g) with an ion exchange capacity of 85 mequ. / 100 g material, suspended in 300 ml warm (60 °C) water under stirring for about 3 hrs. was given slowly to a slightly acidic (HCl, pH 4) solution of 5 g of chitosan (degree of deacetylation ca. 85 %) in 500 ml warm 80 °C) water.

[0027] The resulting precipitate was stirred for some time to complete the reaction, filtered and washed with demineralised water unless no chloride ions were any more detectable. The filtered material, dried at 80 °C gave a very hard, yellowish solid material. The materials are stiff after drying extremely and not anymore soluble in water and can be even machined. An analogy to the bone material is obvious since this material is also a composite material.

[0028] The X-ray diffraction data are shown in Figure 2, indicating that already substantial exfoliation has occurred. Figure 2 shows a XRD scan of the plain montmorillonite and of a chitosan-montmorillonite nano-composite showing exfoliation of the clay sheets.

#### Example 3:

[0029] Firstly 50 g of clay with a cation exchange capacity (CEC) of 128meq/100g is slowly added to 950 ml of warm water (60°C) while stirring. After approximately 1hr of stirring, the mixture is sonified in preferably a plastic bottle for a minimum of 2 hrs. Then the solution is allowed to swell for at least 24 hrs. 55 g of cationic starch with a degree of substitution, DS, of 0.3 is diluted in 2.5 l demineralised water and acidified with 11.7 ml concentrated HCl.

[0030] The clay suspension is added quickly to mixture while stirring. After approximately 5 minutes, when the reaction is complete, e.g. the precipitate is homogeneous in shape and colour; the precipitate is washed until no chlorine can be detected in the filtrate water in an AgNO<sub>3</sub>-test.

[0031] The precipitate was freeze-dried and used for a second mixing step with native potato starch, water and glycerol in an extruder resulting in a completely transparent solid material. The resulting composite material can be very flexible or very hard depending on the starch/plasticizer/clay ratios. The clay platelet distribution in this material was characterized using SEM and show the exfoliation of the smectic material and an homogeneous incorporation into the matrix starch polymer.

#### Example 4:

[0032] A smectite clay mineral (1 g) with an ion ex-

change capacity of 128 mequ. / 100 g material was dissolved in 80 ml water, heated up to 80 °C and ion exchanged with 0.33 g a,w-aminododecanic acid dissolved in 50 ml acidified (HCl) water. The resulting precipitate was washed with water, freeze-dried and characterized using XRD. In the X-ray diffraction data is seen that complete ion exchange of the sodium ions against the organic cations has occurred (change of d-spacing from 12.4 Å to 17 Å). A coextrusion of the intercalate of the organic cation with the clay and of native potato starch as a matrix material together with plasticizers leads to a thermoplastic composite material with homogeneously dispersed, exfoliated platelets. This has been proved again by XRD, TEM and SEM. All methods show partial up to complete exfoliation of the clay material and a homogeneous incorporation into the matrix polymer.

#### Claims

1. Biodegradable thermoplastic material comprising a-natural polymer, a plasticizer and a clay having a layered structure and a cation exchange capacity of from 30 to 250 milliequivalents per 100 gram.
2. Material according to claim 1, wherein the cation exchange capacity is from 50 to 200 milliequivalents.
3. Material according to claim 1 or 2, wherein the clay is a smectite-like clay mineral.
4. Material according to any of the preceding claims, wherein the clay is present in an amount of from 0.1 to 90 wt.%, based on the weight of the biodegradable thermoplastic material.
5. Material according to any of the preceding claims, wherein the natural polymer is a carbohydrate or a protein.
6. Material according to claim 5, wherein the natural polymer is chosen from the group of starch, cellulose, chitosan, alginic acid, inulin, pectin, and derivatives and combinations thereof.
7. Material according to claim 5, wherein the derivative of the natural polymer has a degree of substitution of at least one, and wherein the material does not comprise a plasticizer.
8. Material according to any of the preceding claims, wherein the plasticizer is water, glycerol, ethylene glycol, an oligomer of ethylene glycol, or a combination thereof.
9. Material according to any of the preceding claims, wherein the plasticizer is present in an amount of

15 to 60 wt.%, based on the weight of the biodegradable thermoplastic material.

10. Article of manufacture of a biodegradable thermoplastic material according to any of the preceding claims. 5
11. Process for preparing a biodegradable thermoplastic material according to claims 1-9, comprising preparing a suspension of the clay in water and/or another plasticizer, and extruding the suspension together with the natural polymer and the plasticizer at elevated temperature. 10
12. Process according to claim 11, wherein the temperature is between 35 and 200°C. 15
13. Process according to claim 11 or 12, wherein the suspension of the clay is heated before extrusion to a temperature of up to 60°C in order to exfoliate the clay. 20
14. Process according to claims 11-13, wherein the clay suspension is ion exchanged with at least part of the natural polymer. 25
15. Process according to claims 11-14, wherein the clay suspension is ion exchanged with a modifying agent, the modifying agent being a surfactant having from 6 to 16 carbon atoms, a functionality compatible with the natural polymer and an ammonium, phosphonium or sulfonium group. 30
16. Use of a clay having a cation exchange capacity of from 30 to 250 milliequivalents per 100 gram for improving the stability of a biodegradable thermoplastic material comprising a natural polymer and a plasticizer. 35

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Figure 1

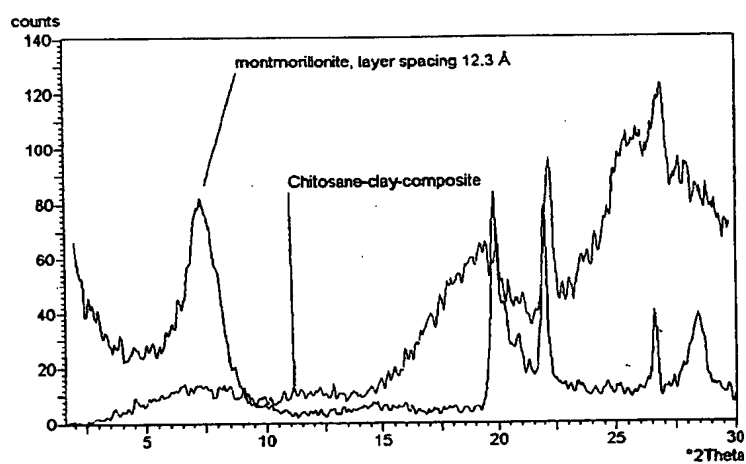


Figure 2



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# EUROPEAN SEARCH REPORT

Application Number  
EP 00 20 0896

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The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 24 July 2000	Examiner Siemens, T
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**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

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